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## The first generation and trapping of a five-membered ring allene: 2-dehydro-3a,4,5,6,6a-pentahydropentalene

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Abstract—3-Bromo-3-fluorotricyclo[ $3.3.0.0^{2,4}$ ]octane (8) was prepared by addition of bromofluorocarbene to bicyclo[3.2.0]hept-6ene (5). Treatment of a solution of 8 in ether with MeLi in the presence of furan afforded the trapping product 11. The formation of the trapping product is consistent with the first generation of a five-membered ring allene, namely, 2-dehydro-3a,4,5,6,6a-pentahydropentalene (4), a reactive intermediate. © 2002 Elsevier Science Ltd. All rights reserved.

Allenes are an important class of unsaturated hydrocarbons, which contain two cumulated double bonds with an orthogonal geometry. Incorporation of an allene unit in small carbocyclic rings causes considerable deformation from the linear geometry and, furthermore, the ring constraints exert torsion towards the planar arrangement of ligands.<sup>1</sup>



Enormous effort has been devoted toward the synthesis of 1,2-cycloheptadiene (1) and 1,2-cyclohexadiene (2), and a number of their derivatives.<sup>1</sup> Among the cyclic allenes, experimental evidence for 1,2-cyclopentadiene (3) has remained elusive.<sup>2</sup> Favorskii first attempted to generate this strained allene by treatment of 1,2-dibro-mocyclopentene with metallic sodium and isolated 1,3-cyclopentadiene.<sup>3</sup> Dehydrobromination reactions of 1-bromocyclopentene<sup>3,4</sup> and 1-chlorocyclopentene<sup>5</sup> resulted in the formation of cyclopentyne rather than the cyclic allene **3**. Fluoride ion-promoted elimination of 2-bromo-3-trimethylsilylcyclopentene<sup>6</sup> gave Wurtz-like dimers<sup>7</sup> instead of the cyclic allene **3**. In this communication, we describe a possible generation of a

five-membered ring allene, namely 2-dehydro-3a,4,5,6,6a-pentahydropentalene (4).

In principle, one of the best ways to generate allenes directly is rearrangement of cyclopropylidenes to cyclic allenes. This method has been successfully applied to the synthesis of six-<sup>8</sup> and seven-membered<sup>9</sup> ring allenes.

For the generation of a five-membered ring allene, the addition of a dihalocarbene to a cyclobutene unit is necessary. Only a few dihalocarbene addition reactions to cyclobutenes are reported in the literature.<sup>10</sup> However, in no case could the addition products be isolated or their structures proven spectroscopically.

First of all, we studied the addition of dibromocarbene<sup>11</sup> generated from CHBr<sub>3</sub> and NaOH under phase-transfer conditions to bicyclo[3.2.0]hept-6-ene  $(5)^{12}$  and isolated the rearranged product 7 in 35% yield (based on recovered starting material) (Scheme 1).

The structure of 7 was elucidated on the basis of NMR spectroscopic data. The stereochemistry of the allylic bromide defines the stereochemistry of the carbene addition. The *exo*-configuration of the bromide implies an *exo*-addition of dibromocarbene to **5**. Since the dibromocarbene adduct **6** undergoes a ring-opening reaction, we decided to add fluorobromocarbene to bicyclo[3.2.0]hept-6-ene (**5**) to prevent the ring-opening reaction at least in one of the isomers. Addition of fluorobromocarbene, generated from CHFBr<sub>2</sub><sup>13</sup> under similar conditions to **5**, afforded the expected addition product, fluorobromocyclopropane derivative **8**, and the ring-opened product **9** in a ratio of 1:5 and in a total yield of 45% (based on recovered starting material) (Scheme 2). The symmetrical structure of the

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Scheme 1.



## Scheme 2.

*exo,exo*-isomer **8** was established by the observation of five signals in its <sup>13</sup>C NMR spectrum, as required by the symmetry in molecule. The initially formed *exo,endo*-isomer **9** underwent a ring-opening reaction to give **10**. The *exo*-orientation of allylic bromide was determined by measuring the coupling constant between protons  $H_{3a}$  and  $H_4$ .

After the successful synthesis and characterization of 8, it was submitted to the Doering-Moore-Skatebol reaction.<sup>14</sup> Treatment of bromofluorocyclopropane 8 with MeLi in ether at -25°C in the presence of furan as the trapping reagent afforded 11 as the major product (20%) and a minor product that was an isomer of 11. The structural assignment of 11<sup>15</sup> was based on <sup>1</sup>H and <sup>13</sup>C NMR data including the 2D NMR spectra COSY, HMQC and HMBC. The *exo*-orientation of proton  $H_2$ in 11 was apparent from the value  $J_{1,2}=4.1$  Hz (dihedral angle=47° determined after AM1 geometry optimization). In the case of the endo-orientation of proton  $H_2$ , the determined dihedral angle would have been 96°. The lack of a measurable coupling between protons  $H_2$ and  $H_3$  and the large coupling observed between the protons  $H_7$  and  $H_8$ ,  $(J_{7.8} = 8.0 \text{ Hz})$  support the exo-orientation of the cyclopentane ring.

In summary, we believe that we have generated the title compound 4, which is the first example of a five-membered cyclic allene, by  $\alpha$ -elimination of Br and F from bromofluoro cyclopropane compound 8 with MeLi.

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- 13-Oxatetracyclo[8.2.1.0<sup>2.9</sup>.0<sup>3,7</sup>]trideca-8,11-diene 11: <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 6.32 (dd, A part of AB system, J<sub>11,12</sub>=5.6, J<sub>10,11</sub>=1.6 Hz, H<sub>11</sub>), 6.17 (dd, B part of AB system, J<sub>11,12</sub>=5.6, J<sub>1,12</sub>=1.5 Hz, H<sub>12</sub>), 5.62 (dd, J<sub>8,7</sub>=8.0, J<sub>8,2</sub>=2.5 Hz, H<sub>8</sub>), 5.20 (bs, H<sub>10</sub>), 5.01 (br. d, J<sub>1,2</sub>=4.1, H<sub>1</sub>), 2.90 (m, H<sub>2</sub>), 2.75 (ddd, J<sub>7,3</sub>=11.0, J<sub>7,8</sub>= 8.0, J<sub>7,6</sub>=1.7 Hz, H<sub>7</sub>), 2.27 (dt, J=4.2 and 11.0 Hz, H<sub>3</sub>), 1.15–1.60 (m, 6H); <sup>13</sup>C NMR δ 146.5 (C9), 136.2 (C11), 130.2 (C12), 117.7 C8), 82.2 (C1), 80.9 (C10), 49.5 (C7), 48.9 (C3), 38.5 (C2), 33.3 (CH<sub>2</sub>), 33.2 (CH<sub>2</sub>), 25.6 (CH<sub>2</sub>); IR (CHCl<sub>3</sub>): 2997, 2934, 2808, 2801, 1220, 1217, 1213, 1208, 785, 780, 773, 770, 760, 750, 737, 669 cm<sup>-1</sup>.